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perfluorinated &lt;and&gt; silica

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**1 Optimum index profile of the perfluorinated polymer-based GI polym optical fiber and its dispersion properties***Ishigure, T.; Koike, Y.; Fleming, J.W.;*

Lightwave Technology, Journal of , Volume: 18 Issue: 2 , Feb 2000

Page(s): 178 -184

[\[Abstract\]](#) [\[PDF Full-Text \(172 KB\)\]](#) **IEEE JNL**

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**2 Status and challenges of GI POF in data-com. area***Koike, Y.; Ishigure, T.;*

Optical Communication, 2001. ECOC '01. 27th European Conference on , Volun 2001

Page(s): 72 -73 vol.1

[\[Abstract\]](#) [\[PDF Full-Text \(376 KB\)\]](#) **IEEE CNF**

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**3 Materials technology for perfluorinated graded-index polymer optical***Blyler, L.L.; White, W.R.; Ratnagiri, R.;*

Optical Communication, 2001. ECOC '01. 27th European Conference on , Volun 2001

Page(s): 64 -65 vol.1

[\[Abstract\]](#) [\[PDF Full-Text \(351 KB\)\]](#) **IEEE CNF**

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**4 Plastic fibers***Yasuhiro, K.;*

Optical Fiber Communication. OFC 97., Conference on , 16-21 Feb 1997

Page(s): 325

[\[Abstract\]](#) [\[PDF Full-Text \(80 KB\)\]](#) **IEEE CNF**

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**5 High-bandwidth, low-loss graded-index polymer optical fiber for near-infrared use**

*Ishigure, T.; Nihei, E.; Koike, Y.;*

Optical Communication, 1998. 24th European Conference on , Volume: 1 , 20-: 1998

Page(s): 231 -232 vol.1

[\[Abstract\]](#) [\[PDF Full-Text \(168 KB\)\]](#) **IEEE CNF**

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**6 POF is overcoming silica in bit rate**

*Ishigure, T.; Koike, Y.;*

Optical Fiber Communication Conference and Exhibit, 2001. OFC 2001 , 2001  
Page(s): ThC7 -T1-3 vol.4

[\[Abstract\]](#) [\[PDF Full-Text \(236 KB\)\]](#) **IEEE CNF**

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**7 Recent status of perfluorinated graded index plastic optical fiber and novel termination method**

*Watanabe, Y.; Onishi, T.; Tsukamoto, T.; Matsuyama, Y.;*

Optical Fiber Communication Conference and Exhibit, 2001. OFC 2001 , 2001  
Page(s): ThC6 -T1-3 vol.4

[\[Abstract\]](#) [\[PDF Full-Text \(248 KB\)\]](#) **IEEE CNF**

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## Perfluorinated Polymer Monolayers on Porous Silica for Materials with Super Liquid Repellent Properties

Jung, D.-H.; Park, I. J.; Choi, Y. K.; Lee, S.-B.; Park, H. S.; Ruhe, J.;  
*Langmuir*: **(Article)**; **2002**; 18(16): 6133-6139. DOI: [10.1021/la025558u](https://doi.org/10.1021/la025558u)

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## Nucleophilic Displacements in Supercritical Carbon Dioxide Using Silica-Supported Phase Agents

DeSimone, J.; Selva, M.; Tundo, P.;  
*J. Org. Chem.*; **(Technical Note)**; **2001**; 66(11); 4047-4049. DOI: [10.1021/jo001337m](https://doi.org/10.1021/jo001337m)

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### Fluorous Synthesis with Fewer Fluorines (Light Fluorous Synthesis): Separation of Tagged and Untagged Products by Solid-Phase Extraction with Fluorous Reverse-Phase Silica Gel

Curran, D. P.; Luo, Z.;  
*J. Am. Chem. Soc.*; **(Article)**; **1999**; 121(39); 9069-9072. DOI: [10.1021/ja991496r](https://doi.org/10.1021/ja991496r)

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### Interaction Forces and Zeta Potentials of Cationic Polyelectrolyte Coated Silica Surfaces and in Ethanol: Effects of Chain Length and Concentration of Perfluorinated Anionic Surfactants on Their Binding to the Surface

McNamee, C. E.; Matsumoto, M.; Hartley, P. G.; Mulvaney, P.; Tsujii, Y.; Nakahara, M.; *Langmuir*; **(Article)**; **2001**; *17*(20): 6220-6227. DOI: [10.1021/la010282w](https://doi.org/10.1021/la010282w)

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*J. Am. Chem. Soc.*, **121** (39), 9069 -9072, 1999. 10.1021/ja991496r S0002-7863(99)01496-1

**Web Release Date:** September 15, 1999

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## **Fluorous Synthesis with Fewer Fluorines (Light Fluorous Synthesis): Separation of Tagged from Untagged Products by Solid-Phase Extraction with Fluorous Reverse-Phase Silica Gel**

**Dennis P. Curran\* and Zhiyong Luo**

*Contribution from the Department of Chemistry and Center for Combinatorial Chemistry, University of Pittsburgh, Pittsburgh, Pennsylvania 15260*

*Received May 6, 1999*

### **Abstract:**

Fluorous synthesis involves tagging an organic substrate with a fluorinated tag for the purposes of separation. To date, techniques of fluorous synthesis have relied on liquid-liquid extractions. This paper applies a simple solid-liquid extraction procedure over fluorous reverse-phase silica gel (silica with a fluorocarbon bonded phase) for use in fluorous synthesis. Four amino acids were tagged on nitrogen with the  $C_9F_{19}CO-$  group, and the resulting acids were coupled in a parallel experiment with an excess of four amines. The resulting 16 crude fluorous amide products were separated from all the coupling reagents and excess amine by two-stage filtration through fluorous silica. In 15 of the 16 cases, the products were isolated in good to excellent yield and purity. All of the products are soluble in organic solvents and none is expected to have any significant solubility in fluorous solvents, so the experiment dramatically illustrates the advantages of the solid-liquid extraction over the liquid-liquid extraction. Future prospects for application of fluorous silica are briefly discussed.

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 AN 1987:555099 CAPLUS  
 DN 107:155099  
 TI Investigation by pyrolysis-gas chromatography of the composition of multicomponent polymeric microheterogeneous systems based on some vinyl monomers  
 AU Shadrina, N. E.; Dmitrenko, A. V.; Pavlova, V. F.; Ivanchev, S. S.  
 CS Plastpolym. Okhta Res. Prod. Assoc., Leningrad, USSR  
 SO Journal of Chromatography (1987), 404(1), 183-95  
 CODEN: JOCRAM; ISSN: 0021-9673  
 DT Journal  
 LA English

L5 ANSWER 1 OF 7 CAPLUS COPYRIGHT 2003 ACS  
 AN 2002:594961 CAPLUS  
 DN 137:142236  
 TI Preparation and use of an impregnating, cleaning fluid based on a polysiloxane network, especially for printing rolls  
 IN Nass, Ruediger; Jonschker, Gerhard  
 PA Nanogate Technologies G.m.b.H., Germany  
 SO PCT Int. Appl., 32 pp.  
 CODEN: PIXXD2  
 DT Patent  
 LA German  
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2002061029	A2	20020808	WO 2002-EP952	20020130
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	DE 10104164	A1	20020822	DE 2001-10104164	20010130
	DE 10106342	A1	20020822	DE 2001-10106342	20010212
	DE 10119825	A1	20021107	DE 2001-10119825	20010423
PRAI	DE 2001-10104164	A	20010130		
	DE 2001-10106342	A	20010212		
	DE 2001-10117138	A	20010405		
	DE 2001-10119825	A	20010423		

L5 ANSWER 2 OF 7 CAPLUS COPYRIGHT 2003 ACS  
 AN 2000:681835 CAPLUS  
 DN 133:366084  
 TI Transformation of MTBE over a solid acid catalyst  
 AU Richards, Sarah A.; Zhang, Wei-xian  
 CS Department of Civil and Environmental Engineering, Lehigh University, Bethlehem, PA, 18015, USA  
 SO Chemical Oxidation and Reactive Barriers: Remediation of Chlorinated and Recalcitrant Compounds, International Conference on Remediation of Chlorinated and Recalcitrant Compounds, 2nd, Monterey, CA, United States, May 22-25, 2000 (2000), 249-255. Editor(s): Wickramanayake, Godage B.; Gavaskar, Arun R.; Chen, Abraham S. C. Publisher: Battelle Press, Columbus, Ohio.  
 CODEN: 69AIJ2  
 DT Conference  
 LA English  
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 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L5 ANSWER 3 OF 7 CAPLUS COPYRIGHT 2003 ACS  
AN 2000:349798 CAPLUS  
DN 133:363157  
TI Inorganic-organic copolymers - materials with a high potential for  
chemical modification  
AU Rose, Klaus; Amberg-Schwab, Sabine; Heinrich, Matthias  
CS Fraunhofer-Institut fur Silicatforschung, Wurzburg, D-97082, Germany  
SO Organosilicon Chemistry IV: From Molecules to Materials, [Lectures and  
Poster Contributions presented at the Muechner Silicontage], 4th, Muechen,  
Apr., 1998 (2000), Meeting Date 1998, 613-619. Editor(s): Auner, Norbert;  
Weis, Johann. Publisher: Wiley-VCH Verlag GmbH, Weinheim, Germany.  
CODEN: 68ZMAL  
DT Conference; General Review  
LA English  
RE.CNT 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L5 ANSWER 4 OF 7 CAPLUS COPYRIGHT 2003 ACS  
AN 1999:99321 CAPLUS  
DN 130:238192  
TI Ceramers based on crosslinked epoxy resins-silica hybrids: low surface  
energy systems  
AU Mascia, L.; Tang, T.  
CS Institute of Polymer Technology and Materials Engineering, Loughborough  
University, Loughborough, LE11 3TU, UK  
SO Journal of Sol-Gel Science and Technology (1998), 13(1/2/3), 405-408  
CODEN: JSGTEC; ISSN: 0928-0707  
PB Kluwer Academic Publishers  
DT Journal  
LA English  
RE.CNT 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD  
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L5 ANSWER 5 OF 7 CAPLUS COPYRIGHT 2003 ACS  
AN 1998:588332 CAPLUS  
DN 129:281577  
TI Schnell gel: rapid formation of low density gels from a  
tetra(fluoroalkoxy)silane  
AU Sharp, Kenneth G.  
CS Central Research, DuPont Co., Wilmington, DE, 19880-0323, USA  
SO Materials Research Society Symposium Proceedings (1998),  
520(Nanostructured Powders and Their Industrial Applications), 123-135  
CODEN: MRSPDH; ISSN: 0272-9172  
PB Materials Research Society  
DT Journal  
LA English  
RE.CNT 24 THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L5 ANSWER 6 OF 7 CAPLUS COPYRIGHT 2003 ACS  
AN 1989:633699 CAPLUS  
DN 111:233699  
TI **Perfluorinated**-ionomer-membrane-based microcomposites. Silicon  
oxide filled membranes  
AU Mauritz, K. A.; Storey, R. F.; Jones, C. K.  
CS Dep. Polym. Sci., Univ. South Mississippi, Hattiesburg, MS, 39406-0076,  
USA  
SO ACS Symposium Series (1989), 395(Multiphase Polym.: Blends Ionomers),  
401-17  
CODEN: ACSMC8; ISSN: 0097-6156  
DT Journal  
LA English

L5 ANSWER 7 OF 7 CAPLUS COPYRIGHT 2003 ACS

AN 1987:555099 CAPLUS  
 DN 107:155099  
 TI Investigation by pyrolysis-gas chromatography of the composition of multicomponent polymeric microheterogeneous systems based on some vinyl monomers  
 AU Shadrina, N. E.; Dmitrenko, A. V.; Pavlova, V. F.; Ivanchev, S. S.  
 CS Plastpolym. Okhta Res. Prod. Assoc., Leningrad, USSR  
 SO Journal of Chromatography (1987), 404(1), 183-95  
 CODEN: JOCRAM; ISSN: 0021-9673  
 DT Journal  
 LA English

L5 ANSWER 3 OF 7 CAPLUS COPYRIGHT 2003 ACS  
 AN 2000:349798 CAPLUS  
 DN 133:363157  
 TI Inorganic-organic copolymers - materials with a high potential for chemical modification  
 AU Rose, Klaus; Amberg-Schwab, Sabine; Heinrich, Matthias  
 CS Fraunhofer-Institut fur Silicatforschung, Wurzburg, D-97082, Germany  
 SO Organosilicon Chemistry IV: From Molecules to Materials, [Lectures and Poster Contributions presented at the Muechner Silicontage], 4th, Muechen, Apr., 1998 (2000), Meeting Date 1998, 613-619. Editor(s): Auner, Norbert; Weis, Johann. Publisher: Wiley-VCH Verlag GmbH, Weinheim, Germany.  
 CODEN: 68ZMAL  
 DT Conference; General Review  
 LA English  
 CC 36-0 (Physical Properties of Synthetic High Polymers)  
 AB A review with 9 refs. The surface properties of coatings derived from inorg.-org. copolymers were adjusted by the proper choice of monomeric organoalkoxysilanes of the general type  $R'_nSi(OR)_{4-n}$  ( $n = 1$  or  $2$ ). Special compds. with functional groups in  $R'$  were incorporated into an inorg. backbone via **hydrolysis** and condensation reactions during sol-gel processing forming an inorg.-org. hybrid material. **Perfluorinated** alkyl chains in  $R'$  reduce the surface energy, thus facilitating anti-adhesive behavior of the resulting coating against polar and nonpolar substances. Due to the presence of ionic compds., e.g. ammonium moieties, the sp. surface resistance is decreased from 1015 to 108 .OMEGA.. Thus elec. charging of the surface is inhibited and the attraction of dust particles is avoided. For a special application in sensor technol. a polyacryloxysiloxane based coating modified with secondary amines is used as a CO<sub>2</sub>-sensitive layer on **silica** optical fibers. The reaction of amino groups with CO<sub>2</sub> can be detected by optical means.  
 ST review inorg org polymer chem modification  
 IT Polysiloxanes, miscellaneous  
 Polysiloxanes, miscellaneous  
 RL: MSC (Miscellaneous)  
 (fluorine-contg.; inorg.-org. copolymers with high potential for chem. modification)  
 IT Fluoropolymers, miscellaneous  
 Fluoropolymers, miscellaneous  
 RL: MSC (Miscellaneous)  
 (polysiloxane-; inorg.-org. copolymers with high potential for chem. modification)  
 IT Hybrid organic-inorganic materials  
 (siloxane-based; inorg.-org. copolymers with high potential for chem. modification)  
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(5) Novak, B; Adv Mater 1993, V5, P6  
(6) Owen, M; Ind Eng Chem Prod Res Dev 1980, V19, P97 CAPLUS  
(7) Rose, K; J Sol-Gel Sci Technol 1998, V13, P729 CAPLUS  
(8) Schubert, U; Chem Mater 1995, V7, P2010 CAPLUS  
(9) Yoshida, Y; Chem Lett 1984, P1571 CAPLUS

L5 ANSWER 5 OF 7 CAPLUS COPYRIGHT 2003 ACS  
AN 1998:588332 CAPLUS  
DN 129:281577  
TI Schnell gel: rapid formation of low density gels from a  
tetra(fluoroalkoxy)silane  
AU Sharp, Kenneth G.  
CS Central Research, DuPont Co., Wilmington, DE, 19880-0323, USA  
SO Materials Research Society Symposium Proceedings (1998),  
520(Nanostructured Powders and Their Industrial Applications), 123-135  
CODEN: MRSPDH; ISSN: 0272-9172  
PB Materials Research Society  
DT Journal  
LA English  
CC 66-4 (Surface Chemistry and Colloids)  
Section cross-reference(s): 78

AB A new family of simple precursors to **silica** gel has been  
developed. The gel precursors are tetra(polyfluoroalkoxy)silanes, the  
prototype being Si(OCH<sub>2</sub>CF<sub>3</sub>)<sub>4</sub>. Formation of transparent monolithic gels  
with no added catalyst can be six orders of magnitude faster than  
comparable reactions of Si(OCH<sub>2</sub>CH<sub>3</sub>)<sub>4</sub> [TEOS]. Extremely low d. gels can be  
generated in minutes at concns. at which TEOS does not gel at all. Pore  
sizes in the wet gels were estd. from hydrodynamic relaxation in a  
beam-bending expt. on cylindrical logs. In a gel at 1% solids, the pore  
size was approx. 100 nm. Monolithic gels can be created at concns. at  
least as low as 0.1% solids and have higher moduli than predicted. NMR  
and GC/IR evidence indicates extremely facile **hydrolysis** and  
condensation pathways and very few silanol or cyclic intermediates in the  
sol. The chem. can also be conducted in **perfluorinated**  
solvents, enabling synthesis of **silica**/fluoropolymer  
nanocomposites.

ST tetrafluoroethoxysilane prepn gelation hydrolysis condensation  
IT Gelation  
Sol-gel transition  
(rapid formation of low d. gels from tetra(fluoroalkoxy)silane)  
IT Condensation reaction  
Hydrolysis  
Pore size  
Xerogels  
(rapid formation of low d. gels from tetra(fluoroalkoxy)silane in  
relation to)

IT 338-39-6P, Silicic acid (H<sub>4</sub>SiO<sub>4</sub>), tetrakis(2,2,2-trifluoroethyl) ester  
RL: PEP (Physical, engineering or chemical process); PNU (Preparation,  
unclassified); RCT (Reactant); PREP (Preparation); PROC (Process); RACT  
(Reactant or reagent)  
(rapid formation of low d. gels from tetra(fluoroalkoxy)silane)

IT 75-89-8, 2,2,2-Trifluoroethanol 10026-04-7, Silicon tetrachloride  
RL: RCT (Reactant); RACT (Reactant or reagent)  
(rapid formation of low d. gels from tetra(fluoroalkoxy)silane)

IT 597-52-4, Triethylsilanol  
RL: RCT (Reactant); RACT (Reactant or reagent)  
(rapid formation of low d. gels from tetra(fluoroalkoxy)silane in  
relation to)

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